

Section 111 Describing Chemical Reactions Answer Key



Section 111 Describing Chemical Reactions

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Section 11.1 Describing Chemical Reactions. Name Date Class . CHEMICAL REACTIONS. Practice Problems. In your notebook, solve the following problems. Use the 3-step problem-solving approach you learned in Chapter 1. SECTION 11.1 DESCRIBING CHEMICAL REACTIONS. 1. Write the skeleton equation for the reaction between hydrogen and oxygen that produces water. 2.

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In a chemical reaction, one or more reactants change into one or more products. Figure 11.1a shows the ingredients for making leavened bread— flour, salt, yeast, and water. Chemical reactions take place when the ingredients are mixed together and heated in the oven. Figure 11.1b shows the product—a loaf of bread.

11.1 Describing Chemical Reactions 11

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The production of a new substance, a gas, is evidence of a chemical change. Two new substances are produced in this reaction, oxygen gas and liquid water. You could describe this reaction by saying, "Hydrogen peroxide decomposes to form water and oxygen gas." You could also write a word equation.

11.1 Describing Chemical Reactions 11

SECTION 11.1 DESCRIBING CHEMICAL REACTIONS (pages 321–329) This section explains how to write equations describing chemical reactions using appropriate symbols. It also describes how to write balanced chemical equations when given the names or formulas of the reactants and products in a chemical reaction. Writing Chemical Equations (pages 321–323) 1.

SECTION 11.1 DESCRIBING CHEMICAL REACTIONS (pages 321–329)

Use the 3-step problem-solving approach you learned in Chapter 1. SECTION 11.1 DESCRIBING CHEMICAL REACTIONS. 1. Write the skeleton equation for the reaction between hydrogen and oxygen that produces water. 2. Write the skeleton equation for the reaction that produces iron(II) sulfide from iron and sulfur.

SECTION 11.1 DESCRIBING CHEMICAL REACTIONS

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a representation of a chemical reaction; the formulas of the reactants (on the left) are connected by

an arrow with the formulas of the products (on the right). Write the formulas of the reactants to the left of the yields sign (arrow) and the formulas of the products to the right. first write the skeleton equation.

Chapter 11.1 Describing Chemical Reactions Flashcards ...

WS 11.1 Describing Chemical Reactions Mrs Rond 2. 17. Fill in this table. + $\frac{1}{4}$ Used in place of $\frac{1}{4}$ for reversible reactions (s) (l) (g) (aq) Indicates that heat is supplied to the reaction. A formula written above or below the yield sign indicates its use as a catalyst (in this example platinum)

College Chemistry Name WS on Section 11.1 Describing ...

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Section 11.1 Assessment. Describe the steps in writing a balanced chemical equation. Write the skeleton equation for the following reactions: Heating solid copper(II)sulfide in the presence of oxygen gas produces pure copper and sulfur dioxide gas. Iron metal and chlorine gas react to form solid iron(III)chloride.

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SECTION 11.1 DESCRIBING CHEMICAL REACTIONS (pages 321-329)

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Section 8-1 Describing Chemical Reactions 1 8-1 Describing Chemical Reactions Chemical Equation - identifies and relates the amounts of reactants and products in a chemical reaction Some Indicators of a Chemical Reaction 1) Heat and light have been released 2) A gas is produced (you see bubbles) 1

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Chapter 11 ... 1.

Chapter 11

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24 Section 11.1 Describing Chemical Reactions Worksheet ...

c. a symbolic way of describing a chemical reaction d. substances that undergo chemical change e. a chemical equation that does not indicate the ... SECTION 11.1 DESCRIBING CHEMICAL REACTIONS 1. Write the skeleton equation for the reaction between hydrogen and oxygen that produces water.

2.

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